

Synthesis and Comparison of the Magnetic Properties of Iron Sulfide Spinel and Iron Oxide Spinel Nanocrystals

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 Supporting Information

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The iron–sulfur system contains a number of phases possessing complex and interesting chemical, magnetic and electronic properties,^{1,2} including Fe₃S₄, which is isostructural to the iron oxide spinel phase Fe₃O₄. Submicrometer Fe₃S₄ nanoparticles occur in nature as magnetosomes in magnetotactic bacteria,³ and as inclusions in geological sediments, where they are significant contributors to chemical remnant magnetization (CRM).^{4,5} Yet the synthesis of monodisperse Fe₃S₄ nanocrystals and the study of their magnetic properties have received relatively little attention compared to well-researched Fe and Fe₃O₄ nanocrystals.^{6–8} This is mainly due to the relatively narrow range of thermal stability of Fe₃S₄ and the tendency for formation of mixtures of iron sulfide phases,¹¹ making the synthesis of phase-pure, crystalline, monodisperse Fe₃S₄ nanocrystals highly challenging.

Nanocrystalline Fe₃S₄ has been synthesized using hydrothermal methods,^{9,10} and monodisperse 2.5–4.5 nm¹¹ and 16 nm¹² Fe₃S₄ nanocrystals have been synthesized successfully from single-source precursors; these latter syntheses involve the preparation of specific single-source iron sulfide-precursors and can involve extended reaction times. In contrast, the approach reported here uses the simple and rapid in situ sulfidization of the decomposition product of iron(II) acetylacetonate (Fe(acac)₂), to form Fe₃S₄ nanocrystals with diameters <10 nm. The flexibility of this method allows the synthesis of monodisperse Fe₃O₄ nanocrystals of the same average diameter as Fe₃S₄ to enable a direct comparison of magnetic properties. Fe₃O₄ nanocrystals are made by omitting the sulfidization step and allowing atmospheric oxidation during work-up. The magnetic properties of the Fe₃S₄ and Fe₃O₄ nanocrystals were studied experimentally, and the results were compared to theoretical models to understand the origin of differences in temperature and size dependent magnetic properties.

In a typical synthesis of Fe₃S₄ nanocrystals, Fe(acac)₂ (0.30 g, 1.2 mmol) and hexadecylamine (HDA) (15 g) were flushed with N₂. The HDA was then melted and degassed by bubbling through N₂ for 30 min, and the solution was heated to 300 °C for 2 h. Elemental sulfur (1.5 equiv) dissolved in oleylamine (OA) (0.06 g, 1.8 mmol sulfur in 3.6 cm³ OA) was injected rapidly, and the reaction mixture was then immediately cooled to room temperature with a water bath. The nanocrystals were isolated by diluting the

reaction mixture with an equal volume of hot ethanol, and separating the precipitate by centrifugation. Repeated cycles of sonication of the precipitate in hot ethanol followed by centrifugation removed HDA residues. Size selective purification was used to concentrate the smaller size fraction (see the Supporting Information for full experimental details).

A low-magnification TEM image of the Fe₃S₄ nanocrystals is shown in Figure 1A, which have an average diameter of 6.5 ± 0.5 nm (inset). SAED and XRD patterns of the samples are shown in Figure SI-1A in the Supporting Information and Figure 1C, respectively; and were both indexed to the characteristic inverse spinel structure of Fe₃S₄. A comparison of the XRD pattern with the literature pattern of Fe_{1–x}S (see Figure SI-2 in the Supporting Information) shows that the as-synthesized Fe₃S₄ nanocrystals contain no substantial impurities. EDS analysis (see Figure SI-3A in the Supporting Information) found an Fe:S ratio of 1:1.36, close to the expected ratio of 1:1.33 for stoichiometric Fe₃S₄. Figure 1B shows a typical low magnification TEM image of monodisperse Fe₃O₄ nanocrystals with an average diameter of 6.0 ± 0.2 nm. SAED and XRD patterns of the nanocrystals are shown in Figure SI-1B in the Supporting Information and Figure 1C, respectively; and were indexed to Fe₃O₄. EDS analysis (see Figure SI-3B in the Supporting Information) found an Fe:O ratio of 1:1.34, as expected for stoichiometric Fe₃O₄.

Magnetic measurements were conducted on both samples (see the Supporting Information for details), with ZFC/FC plots for Fe₃S₄ and Fe₃O₄ nanocrystals shown in Figure 2. The ZFC/FC plot of the Fe₃S₄ nanocrystals (Figure 2A) has an inflection centered at 50 K, indicating a transition to superparamagnetism above this temperature (i.e., average $T_B \approx 50$ K), however the ZFC and FC plots do not coincide, indicating the retention of some degree of magnetic irreversibility, which may be due to a small amount of high coercivity Fe_{1–x}S impurity. A slight decrease in the susceptibility of the FC curve below the blocking temperature indicates some degree of antiferromagnetic order. In contrast, the ZFC/FC plot of the

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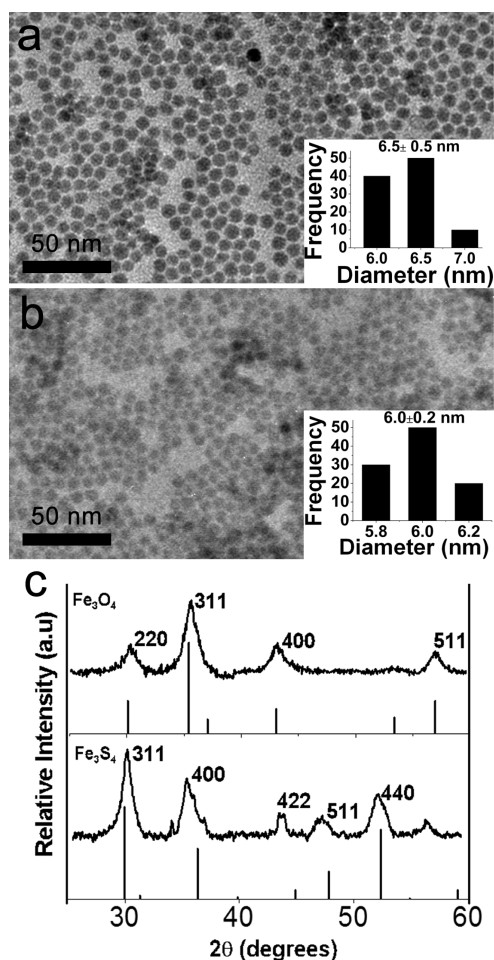


Figure 1. TEM images of (a) Fe_3S_4 nanocrystals, (b) Fe_3O_4 nanocrystals. Insets show histogram of nanocrystal diameters. (c) XRD patterns of Fe_3O_4 and Fe_3S_4 nanocrystals and reference patterns.

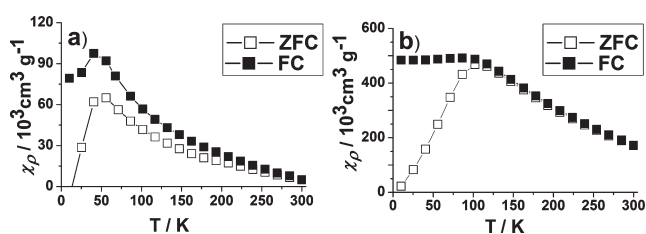


Figure 2. ZFC/FC plots of (a) Fe_3S_4 and (b) Fe_3O_4 nanocrystals.

Fe_3O_4 nanocrystals Figure 2B displays a distinct transition at 100 K, with the ZFC and FC plots coinciding above this temperature, and no decrease in the susceptibility of the FC curve below the blocking temperature. The shape of the FC curves of the samples indicates a degree of interparticle interaction, which can lead to an increase of T_B , when compared to a noninteracting ensemble.¹³

Field loops for both samples at 10 and 300 K are shown in Figure 3. At 10 K, the field loop for Fe_3S_4 nanocrystals Figure 3A displays very low coercivity at low-field, but is very slightly wasp-waisted at moderate fields (with $H_c \approx 140$ Oe), indicating the presence of a high coercivity component either due to larger Fe_3S_4 nanocrystals or Fe_{1-x}S impurity. At both temperatures the field loop for the Fe_3S_4 nanocrystals has a slanted appearance at high field,

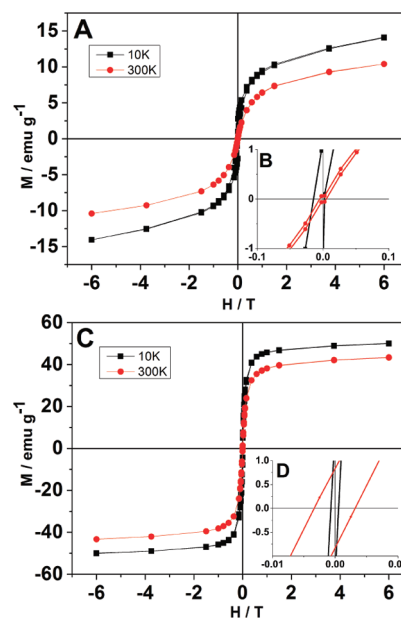


Figure 3. Field loops of (a) Fe_3S_4 nanocrystals at 10 K and 300 K, (b) inset showing hysteretic behavior. (c) Fe_3O_4 nanocrystals at 10 K and 300 K and (d) inset showing hysteretic behavior.

indicating the presence of a paramagnetic component, which is not saturated at 6 T. Because of this, the saturation magnetization, M_s , for the Fe_3S_4 nanocrystals can only be approximated as 12 and 9 emu g^{-1} at 10 and 300 K, respectively. The room temperature saturation magnetization of 9 emu g^{-1} measured for 6.5 nm nanocrystals is larger than that seen for 2.5–4.5 nm Fe_3S_4 nanocrystals (1.5–2.0 emu g^{-1}),¹¹ but is significantly smaller than that reported recently for 30 nm nanoplates (26–31 emu g^{-1}),¹⁴ indicating that the saturation magnetization has a strong size dependence. Field loops for Fe_3O_4 nanocrystals (Figure 3C, D) display little coercivity (≤ 25 Oe) at either 10 or 300 K. At both temperatures, the field loops have square profiles and saturate fully at high field. The Fe_3O_4 nanocrystals have M_s values of 50 and 43 emu g^{-1} at 10 and 300 K, respectively.

To explore the differences in magnetic properties of the two phases, calculations were performed of the energetics and magnetic properties of the bulk spinels, as a function of lattice parameter. DFT calculations use PAW potentials¹⁵ and the PW91¹⁶ functional as implemented in VASP5.2,¹⁷ unless otherwise stated. For the sake of comparison, the LDA¹⁸ functional was also used.

Particle size is known to affect magnetic susceptibility through a number of mechanisms, including alteration of lattice parameter.¹⁹ The variation in magnetic moment with respect to lattice parameter was calculated for Fe_3S_4 and Fe_3O_4 , and is shown in the Supporting Information-Figure SI-4. Observed saturation magnetizations for both bulk and nanocrystalline Fe_3S_4 and Fe_3O_4 , and calculated magnetizations for the energy-minimized models are tabulated in Table 1. The calculated (DFT-PW91) magnetization of the Fe_3O_4 spinel is 112 emu g^{-1} , which is slightly higher than the experimental value for bulk Fe_3O_4 of 91 emu g^{-1} .²⁰ The calculated lattice parameter for the bulk is 8.43 Å, in good agreement with the experimental value of 8.40 Å. The experimental value of 43 emu g^{-1} obtained in this work for the saturation magnetization of 6 nm nanocrystals at 300 K is consistent with previous literature results of 40–42 emu g^{-1} for 7–10 nm Fe_3O_4 nanocrystals.^{19,20} The factor of 2 reductions

Table 1. Magnetic Properties of Fe₃S₄ and Fe₃O₄ Nanocrystals

		M_s (emu g ⁻¹)		T (K)
bulk Fe ₃ S ₄	calcd	43		0
	obsd	38 ²¹		4
bulk Fe ₃ O ₄	calcd	112		0
	obsd	91 ²⁰		4
		M_s (emu g ⁻¹)		T_B (K)
		10 K	300 K	
Fe ₃ S ₄ nanocrystals	obsd	12	9	~50
Fe ₃ O ₄ nanocrystals	obsd	50	43	100

in the magnetization of nanocrystalline Fe₃O₄ relative to the bulk is attributable to the particle size through a combination of surface effects which contribute to weakening of magnetization,¹⁹ and lattice contraction. A contraction of 5–6% in the lattice parameter would give magnetization values in good agreement with the experimentally obtained values here. The actual contraction should be less due to surface effects and a finite temperature.

The calculated lattice parameter for bulk Fe₃S₄ spinel is 9.46 Å, compared to an experimental value of 9.876 Å.²² This is larger than Fe₃O₄, where $a = 8.43$ Å; the agreement with experiment is also poorer, in part due to a very flat energy profile around the minimum. At the calculated energy minimum, the magnetization of the Fe₃S₄ spinel is 43 emu g⁻¹, which corresponds well with the experimental value for the bulk of 38 emu g⁻¹.²⁰ The magnetization would be higher at the experimental lattice parameter, according to GGA-PW91; however, we consider it more appropriate to take ground-state values for comparison with experiment.

LDA calculations performed for comparison found a value of 39 emu g⁻¹ at the same value of $a = 9.46$ Å, consistent with the GGA-PW91 values. These values may be compared with the magnetization values measured in Table 1. The more than 3-fold reduction of magnetization from a calculated 43 emu g⁻¹ in the bulk to an observed 12 emu g⁻¹ at 10 K for nanocrystalline Fe₃S₄ is larger than the 2-fold reduction observed for Fe₃O₄. This indicates that size-related effects,²³ such as lattice contraction and surface magnetic effects play a more significant role for Fe₃S₄ than Fe₃O₄. This may be related to the relatively larger Fe–Fe distance in Fe₃S₄, because of the larger lattice parameter (more than 1 Å larger than Fe₃O₄).

T_B for 6 nm Fe₃S₄ and Fe₃O₄ nanocrystals is 50 and 100 K, respectively. Given the almost identical size distributions of the nanocrystals, the difference in T_B originates in a difference in magnetic anisotropy energy. Magnetic anisotropy energy can have contributions from shape and/or magnetocrystalline anisotropy.²⁴ The effective anisotropy of an ensemble of magnetic nanocrystals, K_{eff} , can be estimated from $K_{\text{eff}} = 25k_B T_B / \langle V \rangle$,²⁵ where k_B is the Boltzmann constant and $\langle V \rangle$ is the average particle volume. Assuming spherical geometries for the nanocrystals and substituting the average radii, K_{eff} values of 16 and -38.5 kJ m⁻³ for Fe₃S₄ and Fe₃O₄, are obtained, respectively. The sign of K_{eff} indicates the orientation of the easy axes of the phase: both Fe₃S₄ and Fe₃O₄ display cubic anisotropy, however the easy axes of Fe₃S₄ lie along the $\langle 100 \rangle$ directions,²⁴ and for Fe₃O₄ lie along the $\langle 111 \rangle$ directions.²⁶ The values obtained may be compared to values from the literature for the bulk phases.

The first-order magnetocrystalline anisotropy constant, K_1 , for Fe₃O₄ has been established as -20 kJ m⁻³ at 4 K,²⁶ but to the best of our knowledge K_1 has only ever been estimated for Fe₃S₄ ($K_1 \approx 1$ kJ m⁻³) and this estimate did not appear to be based on experimental data.²⁷ The deviation of K_{eff} from K_1 for Fe₃O₄ (-20 cf. -38.5 kJ m⁻³) indicates a strong shape anisotropy contribution to the thermal blocking behavior of the Fe₃O₄ nanocrystals. As both sets of nanocrystals have identical morphologies, it can be expected that shape anisotropy contributes a similar proportion to the effective anisotropy of the Fe₃S₄ nanocrystals. This leads to an estimated magnetocrystalline anisotropy constant of 8 kJ m⁻³ for Fe₃S₄.

In conclusion, monodisperse 6 nm Fe₃S₄ spinel nanocrystals have been synthesized by a new sulfidization method, a modification of which enabled the synthesis of isostructural Fe₃O₄ nanocrystals of identical average size. The temperature- and field-dependent magnetic properties of both sets of nanocrystals showed that the Fe₃S₄ nanocrystals had a saturation magnetization of 12 and 9 emu g⁻¹ at 10 and 300 K, respectively, which is a more than 3-fold reduction compared to the calculated bulk value (43 emu g⁻¹). The blocking temperature of the Fe₃S₄ nanocrystals was lower than that of the Fe₃O₄ nanocrystals (50 and 100 K, respectively) and led to an estimate of 8 kJ m⁻³ for the magnetocrystalline anisotropy of the Fe₃S₄ spinel phase.

■ ASSOCIATED CONTENT

S Supporting Information. Details of size selective purification and magnetic measurements. SAED, XRD patterns, and EDS analysis of the Fe₃S₄ and Fe₃O₄ nanocrystals and plot of variation in magnetic moments with lattice parameter. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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